

## Homework solutions

### The goal of physical chemistry - Chapter 1

**Homework Problem 01.1:** For the function  $y=t$ , calculate the average  $\bar{y}$ , if  $P(t)=ke^{-kt}$  per unit time over the range  $t=0$  to  $\infty$ . Since  $t$  is time, this gives the average time for a molecule with rate coefficient  $k$  to decay. How does that compare with the half-life of the molecule? [Hint:  $y=t$  is a continuous function, so you need to use the integral formula, not the summation formula.]

**Solution:** Use the formula  $\bar{y} = \int dx y(x)P(x)$ , where our independent variable 'x' is now called 't' and  $y(x)$  is thus  $y(t)$ :

$$\bar{y} = \int_0^{\infty} dt t k e^{-kt}.$$

You can do this by integration by parts, or just go to an online integrator such as Wolfram Alpha: <https://www.wolframalpha.com/calculators/integral-calculator/> Try it out right now by typing the formula  $x * k * \exp(-k*x)$  into the integral and hitting equal on the right. You get

$$-\frac{e^{-kx}(1+kx)}{k}$$

Evaluating at  $\infty$ , where  $e^{-kt}(1+kt)$  goes to 0, and evaluating at  $t=0$ , where we get  $-(1/k)$ , the result is

$$\bar{y} = 0 - \left(-\frac{1}{k}\right) = \frac{1}{k} = \bar{t}.$$

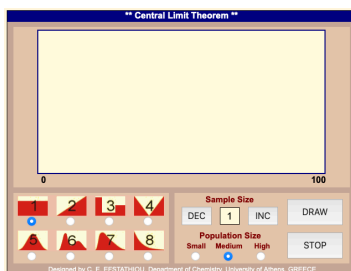
Thus the average time for the reaction is the inverse of the rate coefficient  $k$ , and since the half-life  $t_{1/2} = \ln(2)/k$ , the average time of reaction  $\bar{t} = t_{1/2}/\ln(2)$ .

Why is  $P(t)=ke^{-kt}$  multiplied by the  $k$  out front, and not just  $P(t)=e^{-kt}$ ? The reason is that the probability must be normalized to 1, i.e. the chance the reaction occurs at some time between  $t=0$  and  $\infty$  must be 100% (or 1). But  $\int_0^{\infty} e^{-kt} dt = 1/k$ , not 1, so we multiply by  $k$  to normalize the probability.

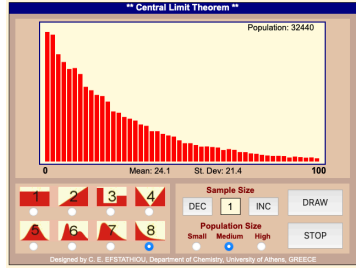
**Homework Problem 01.2:** Go to the applet

[http://195.134.76.37/applets/AppletCentralLimit/Applet\\_CentralLimit2.html](http://195.134.76.37/applets/AppletCentralLimit/Applet_CentralLimit2.html), which shows you how starting with 1 random variable  $X_1$  with distribution  $P(X_1)$ , then adding a second, and so forth, eventually produces a Gaussian. Click on "8" = exponential decay, Population size 'Medium', and draw for sample size 1. Then INCrease sample size to 2 (two random variables, both exponentially distributed added up), 5 and 10. Does it begin to look like a Gaussian for 10 random exponential variables added up?"

**Solution:** Going to the website, you'll see a window



Let's show the first example here, a single random variable, but don't forget to turn in the results for the sum of 2, 5 and 10 random variables as well for full credit! Click on "8" to select an exponential decay  $P(t)=ke^{-kt}$  and then click "DRAW". You'll get, with an x-axis going from 0 to 100



the expected exponential distribution. But if you set "Sample Size" to 2, 5 or 10, the distribution will lose probability at 0 and start looking like a Gaussian. The reason is that while getting "0" is the most likely value to get (highest probability  $P(t)$ ), getting "0" twice in a row is still not as likely as getting, say  $10=0+10$  or  $1+9$  or  $2+8$  or lots of other possibilities. By the time you add up 10 random variables, the chance of getting "0" ten times in a row is negligible, and you get a near-Gaussian centered at the average, about 24 in the online app.

**Homework problem 01.3:** Consider the molecular DNA sequence CCAGCAC\_GGCAGCAT, with a base missing. What would be a good guess for that base? To start, what is the probability  $P(A)$  that adenine occurs in that known sequence? The probability  $P(C)$ ? The probability that whenever there is an A, there's a C before it? (that is  $P(C, \text{ given } A \text{ is in the next position})$ ). Therefore what is  $P(A|C) = P(A, \text{ given } C \text{ is in the previous position})$ ? So what odds do you give an A being in the blank space?

Solution: The known sequence has 15 entries and there are 4 adenines, so  $P(A)=4/15$ . Similarly,  $P(C)=6/15$ . The probability there is a C at position N, given there is an A at position N+1 is  $P(C|A)=1$  (100%). Thus

$$P(A|C)=P(C|A) P(A)/P(C) = 1*(4/15)/(6/15) = 2/3.$$

For guanine = G, the answer would have been

$$P(G|C)=P(C|G) P(G)/P(C) = 0*(4/15)/(6/15) = 0$$

since there is no prior occurrence of a G after a C in the sequence. So A would be a good guess, but Bayes is not giving it a '1': there are extra Cs in the sequence, such as the very first one, that do not have an A after them. If the actual answer came out differently, you can recalculate the probabilities, and the one for A will come out smaller.

Note that such techniques are used to reconstruct phylogenetic trees (ancient DNA sequences). So take the answers with a grain of salt: they are probabilistically reasonable, but by no means guaranteed right.

**Homework problem 01.4:** Write the number  $i$  in exponential notation (what is its angle  $\varphi$ ?). Now take the natural logarithm  $\ln(i)$ . What do you get? Can you devise a simple procedure that extracts the magnitude ' $r$ ' and angle ' $\varphi$ ' of any complex number by distributing them into a real and imaginary piece?

**Solution:** Since  $x=r[\cos(\varphi)+i\sin(\varphi)]$ , we must have  $r=1$  because the magnitude of  $|i|=1$ .  $\cos(\varphi)=0$  must also be true, since  $i$  has no real part. Thus  $\cos(\varphi)=90^\circ$  or  $\pi/2$ . We therefore have

$$i=e^{i\pi/2}$$

in exponential notation. Taking the natural log, we have

$$\ln(i)=\ln(e^{i\pi/2})=(i\pi/2)\ln(e)=i\pi/2$$

So, the log can extract the magnitude and phase angle of a number as follows: if  $x=re^{i\varphi}$ , then  $\ln(x)=\ln(r)+i\varphi$ . The real part will be the log of the magnitude, and the imaginary part will be the angle.